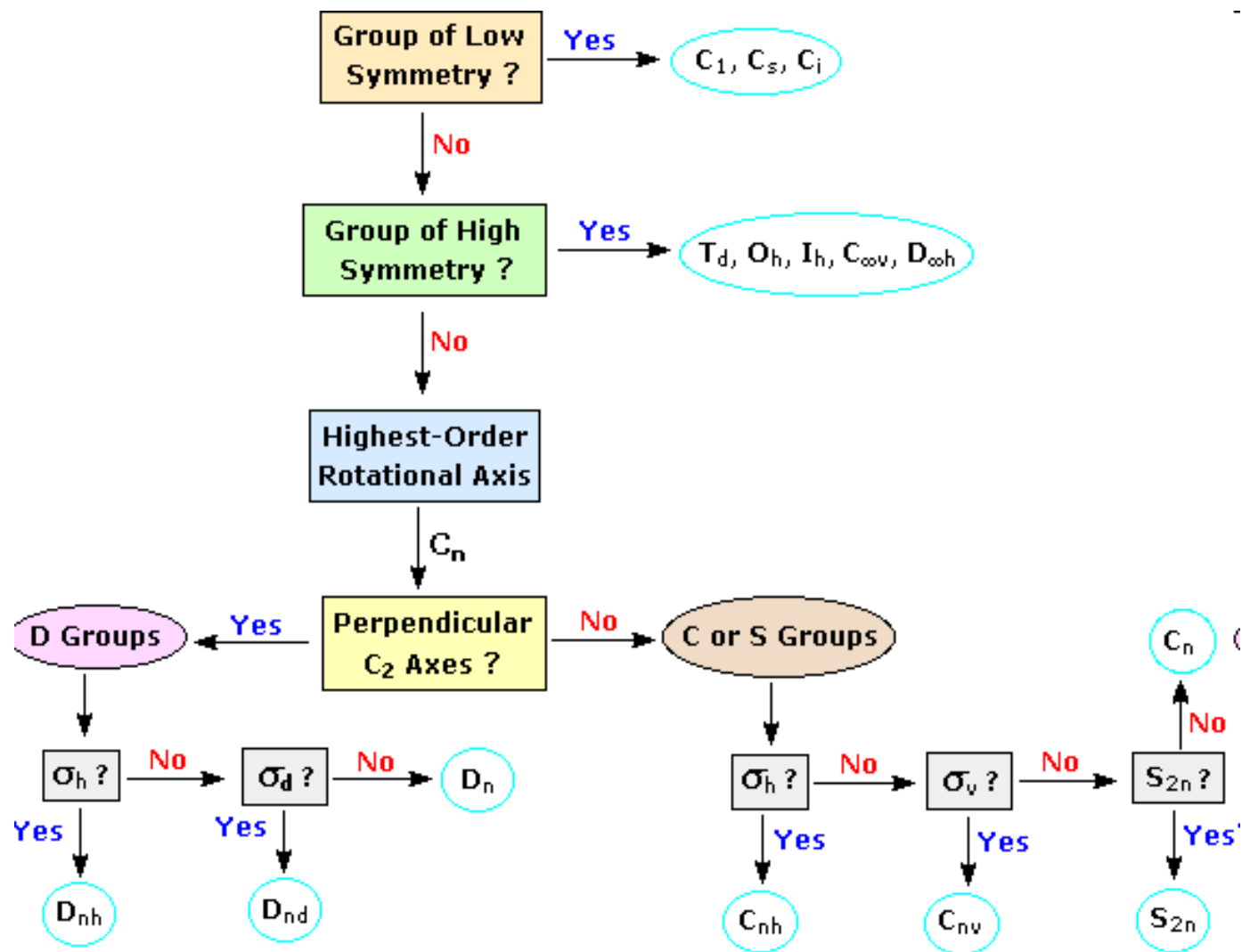


CAPVT VII

SIMETRIJA

Ob simetriji

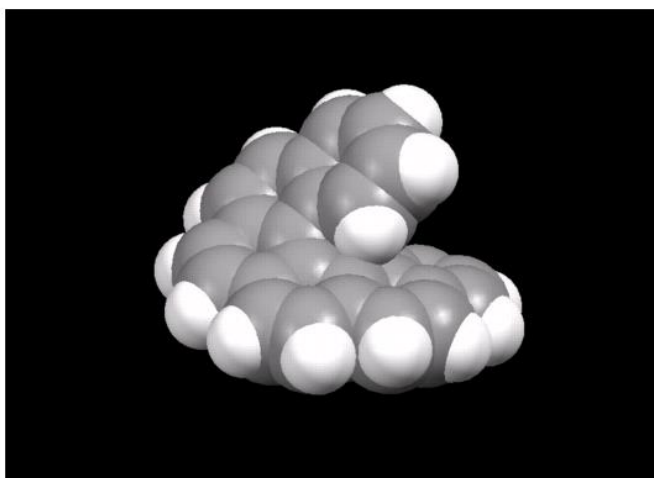
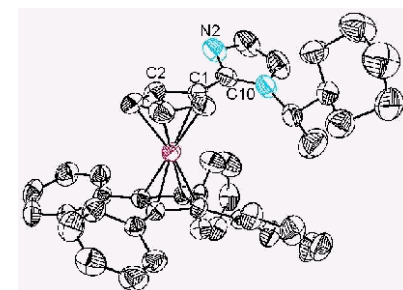
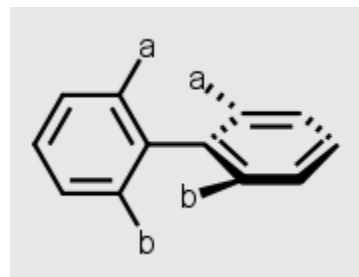
- Elementi simetrije
- Operacije simetrije
- Grupe točke



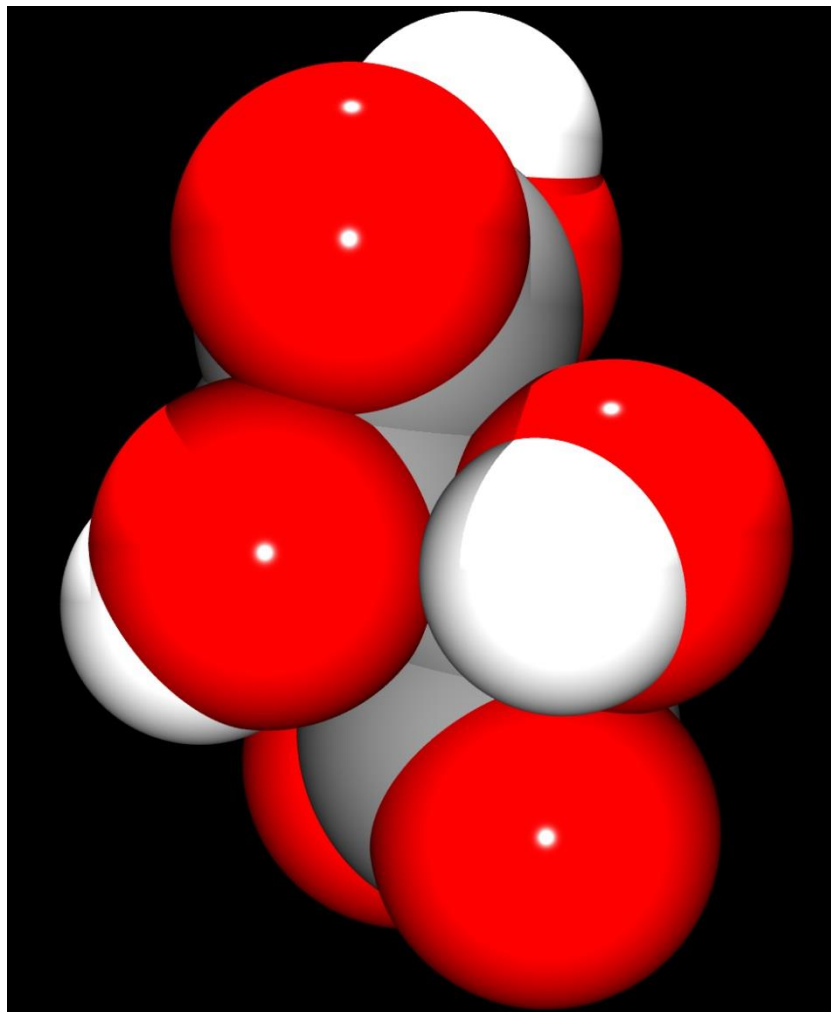
- Kiralnost
 - Izostanak elemenata simetrije 2. vrste
- Polarnost
 - Barem jedna polarna os

Vrste kiralnosti

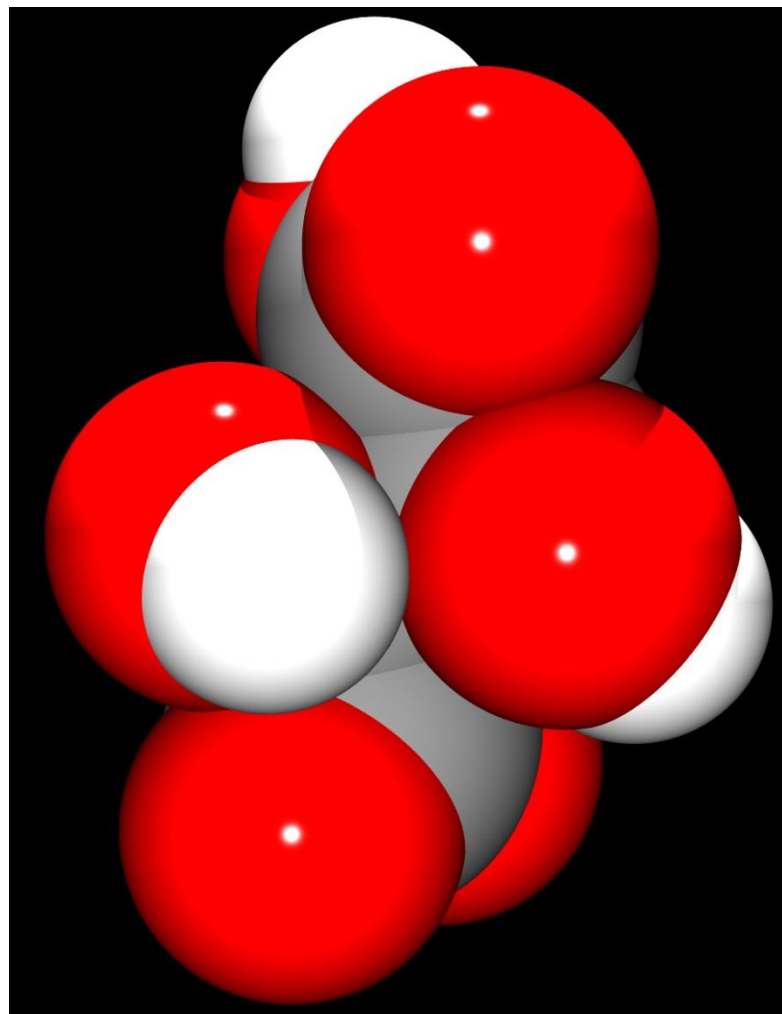
- Centralna
- Osna
- Ravninska
- Helikalna



Lijeva i desna vinska kiselina



(+), L, *R,R*



(-), D, *S,S*

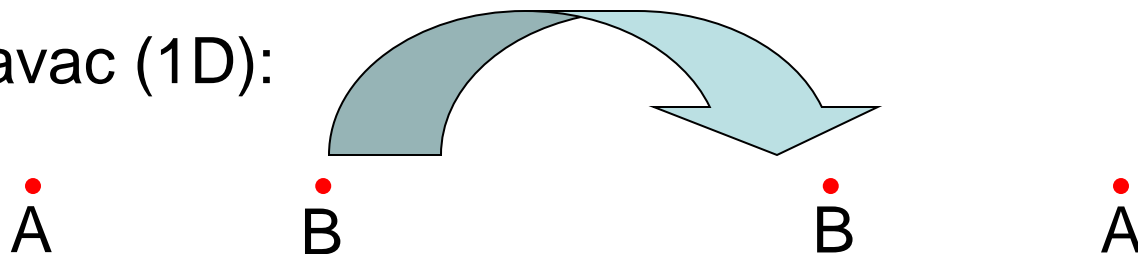
Pasteur, 1847.

- (+), (–) – smjer optičkog zakretanja
- D, L – za ugljikohidrate i aminokiseline... i spojeve koje se može izvesti iz gliceraldehida
- *R*, *S* – za četverovalentne (tetraedarske) atome
- Λ , Δ i λ , δ – za šesterovalentne (oktaedarske) atome
- *M*, *P* – za zavojnice
- Nema univerzalnog sustava!!

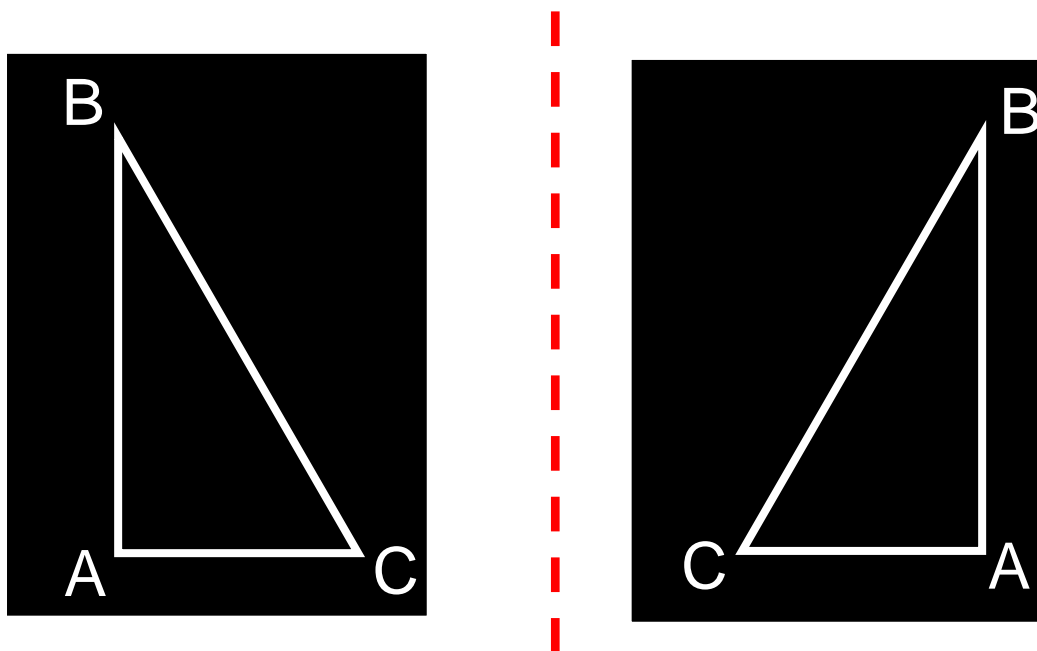
Kako definirati kiralnost?

(Koliko točaka nam za to treba?)

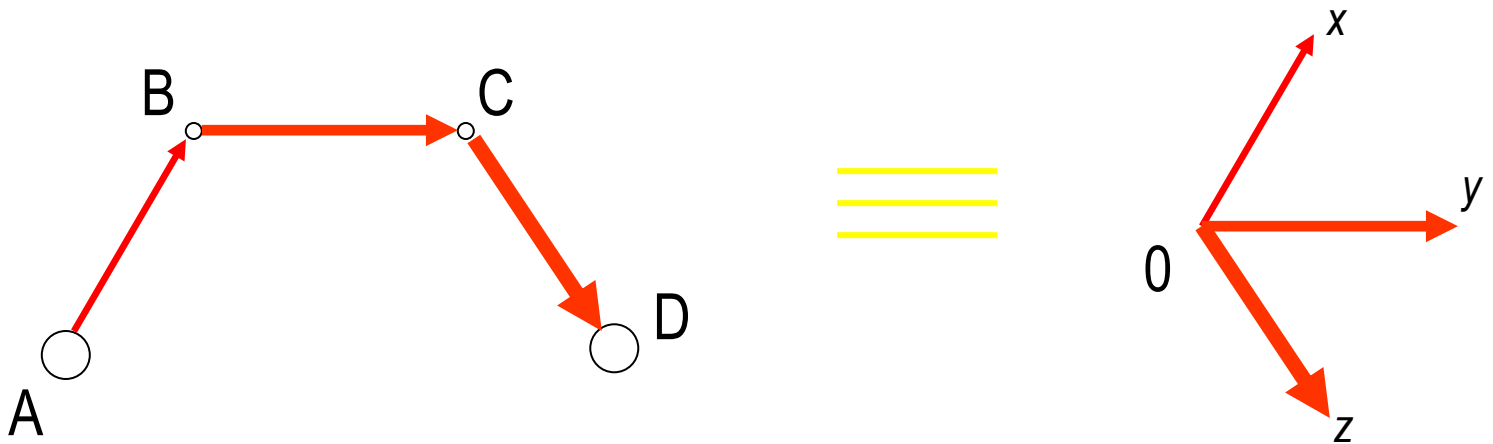
Pravac (1D):



Ravnina:



3D prostor



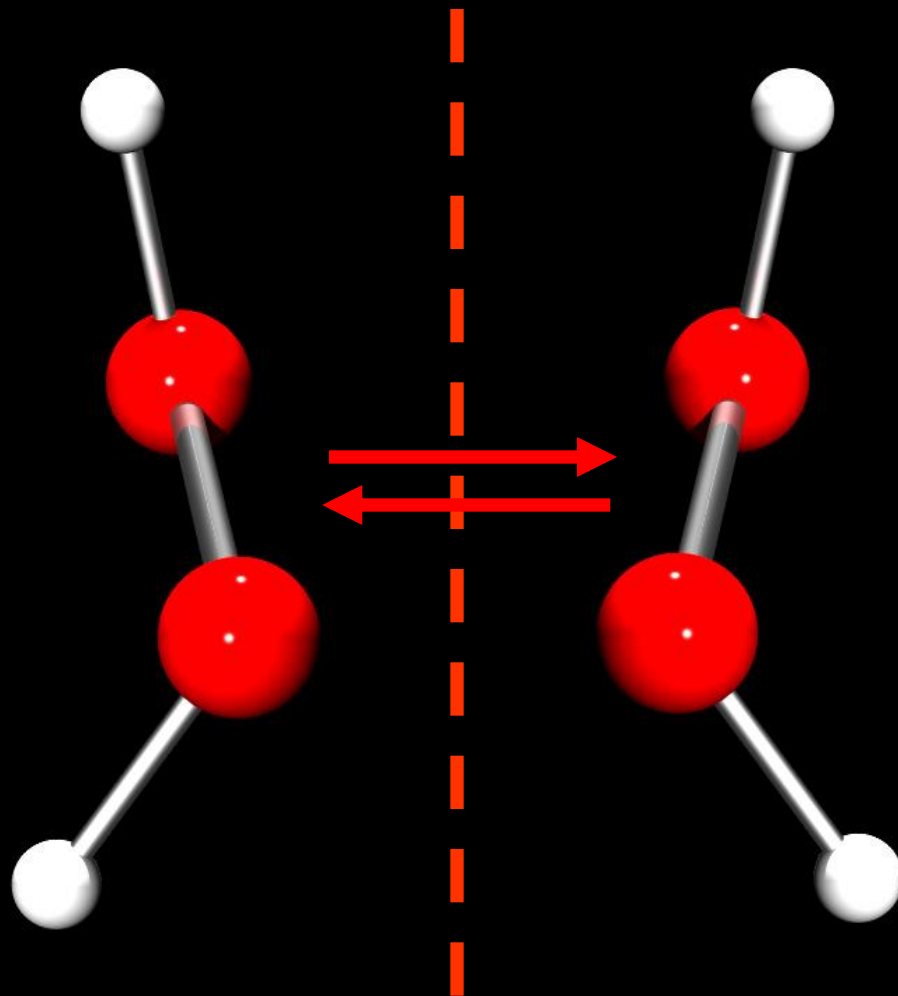
S četiri točke definiramo koordinatni sustav

Koordinatni sustav može biti lijevi i desni

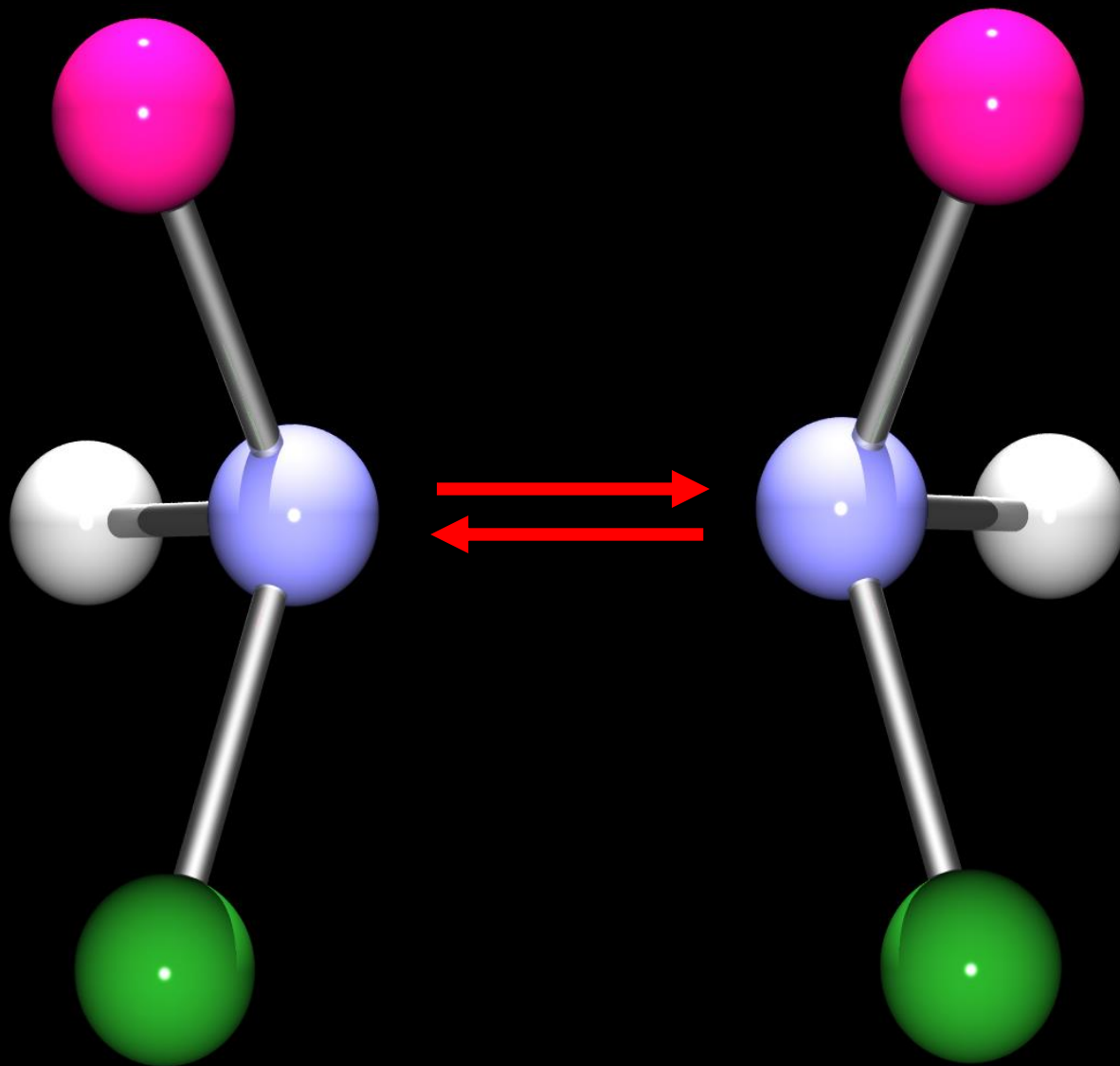
Kiralna molekula mora imati najmanje 4 atoma

Najmanja kiralna molekula?

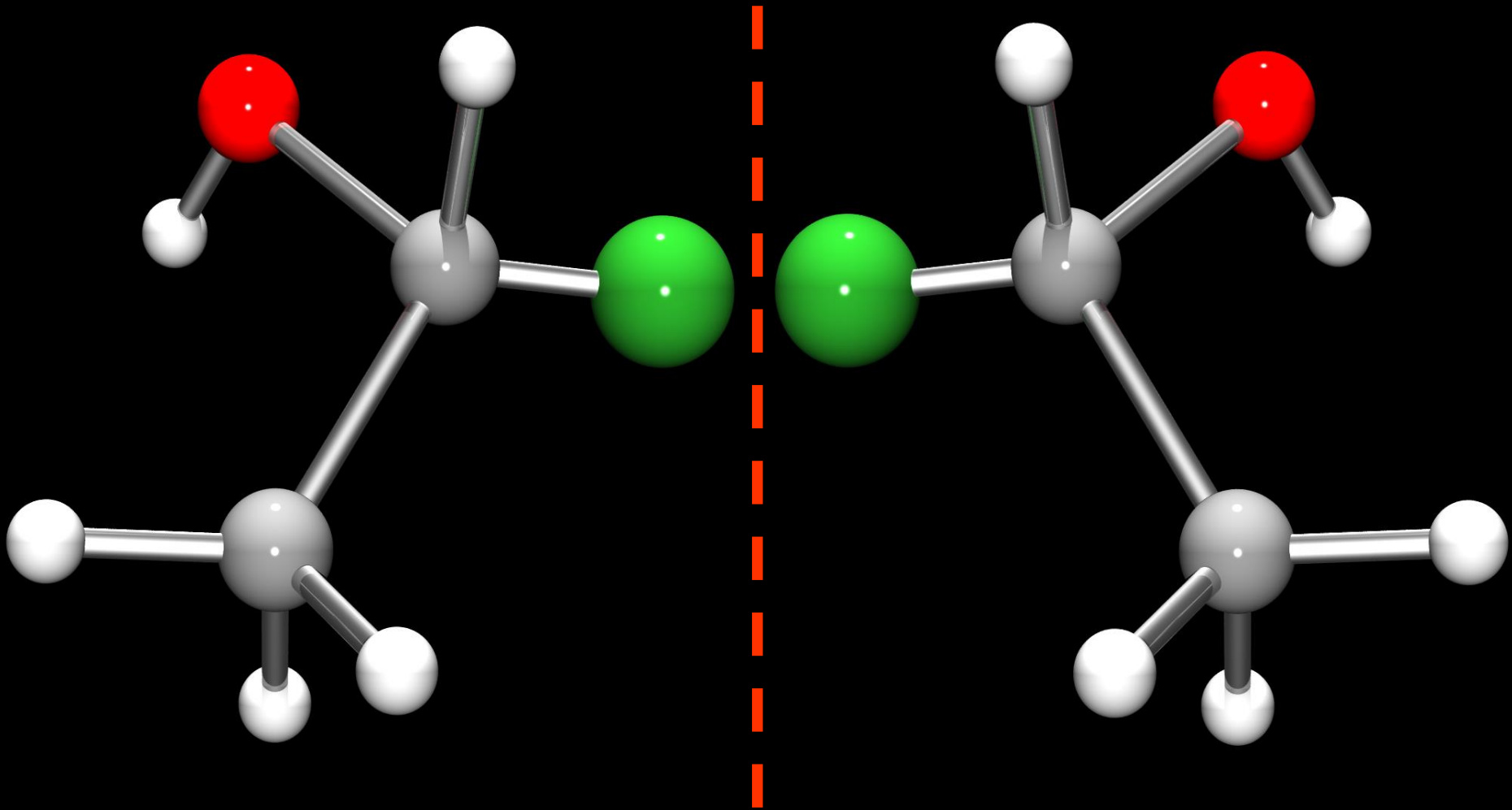
Ima samo 4 atoma:



Tercijarni amin



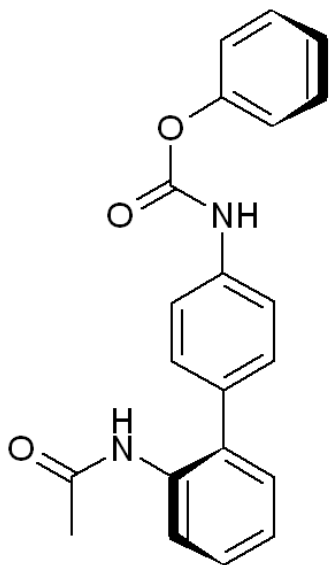
Četverovalentni ugljik: 2 različita spoja!



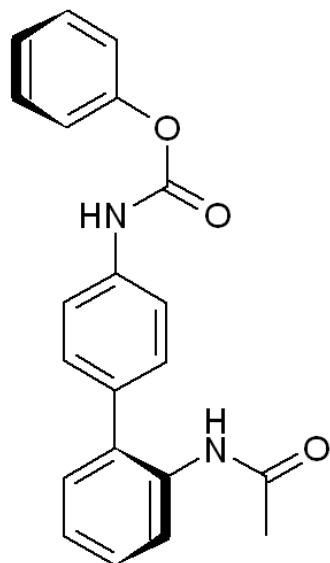
Sterički onemogućena pretvorba lijevog u desno

- Kiralan spoj **ne mora** imati stereogeni centar
 - spojevi sa stereogenim centrom su specijalni slučaj kiralnih spojeva
 - tetraedarski stereogeni centri su specijalni slučaj kiralnih spojeva sa stereogenim centrima
- Molekula je akiralna samo ako ima **centar inverzije** ili **zrcalnu ravninu**
- Ako nema ni centar inverzije ni zrcalnu ravninu, kiralna je!

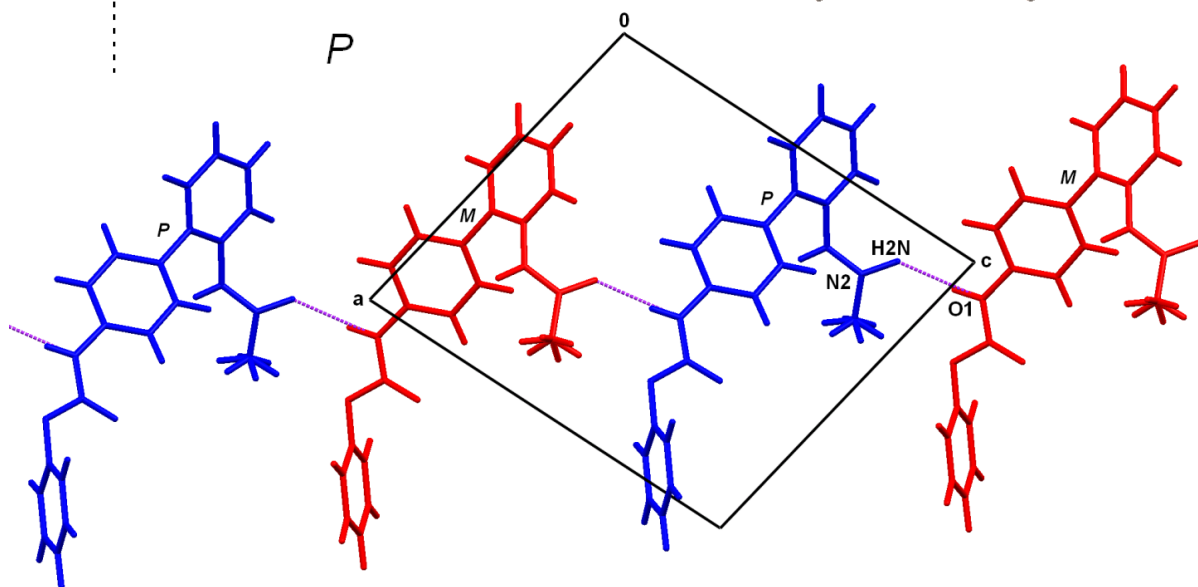
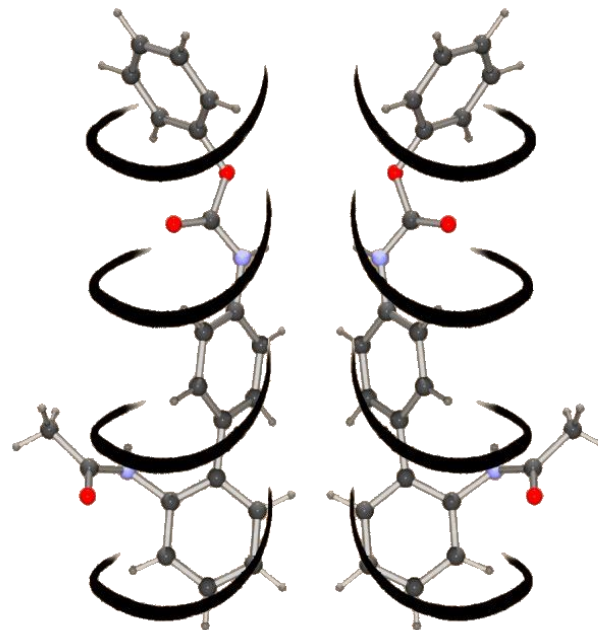
Heličnost



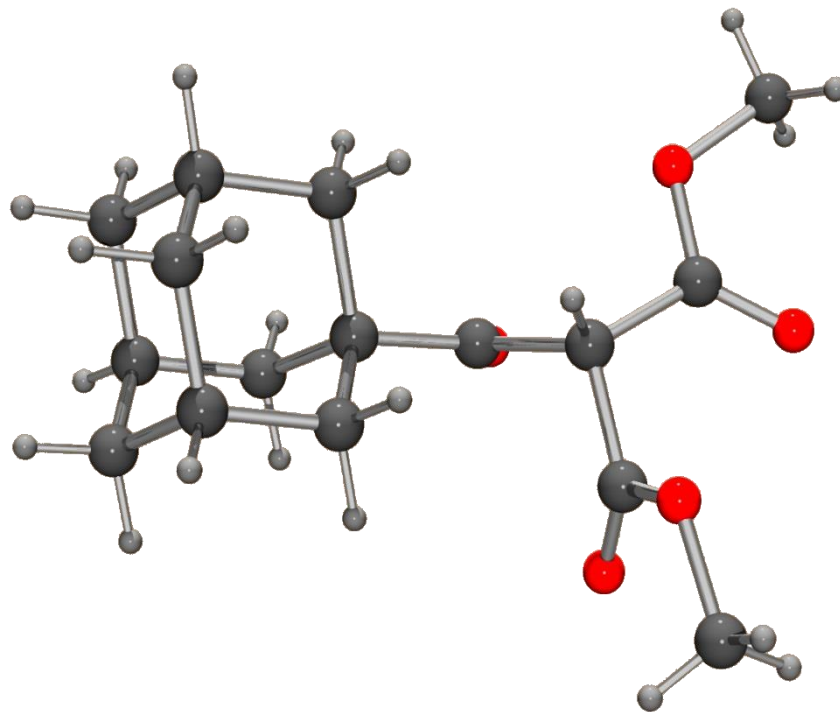
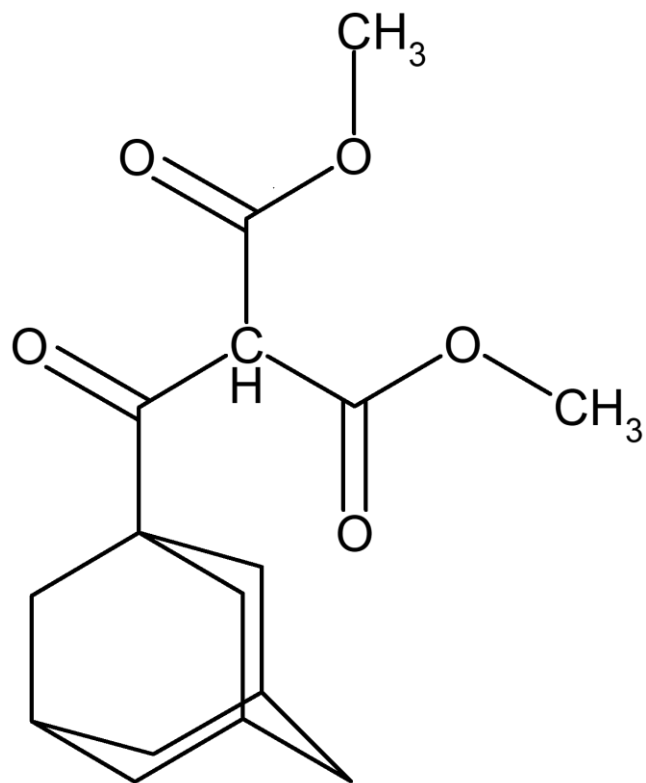
M



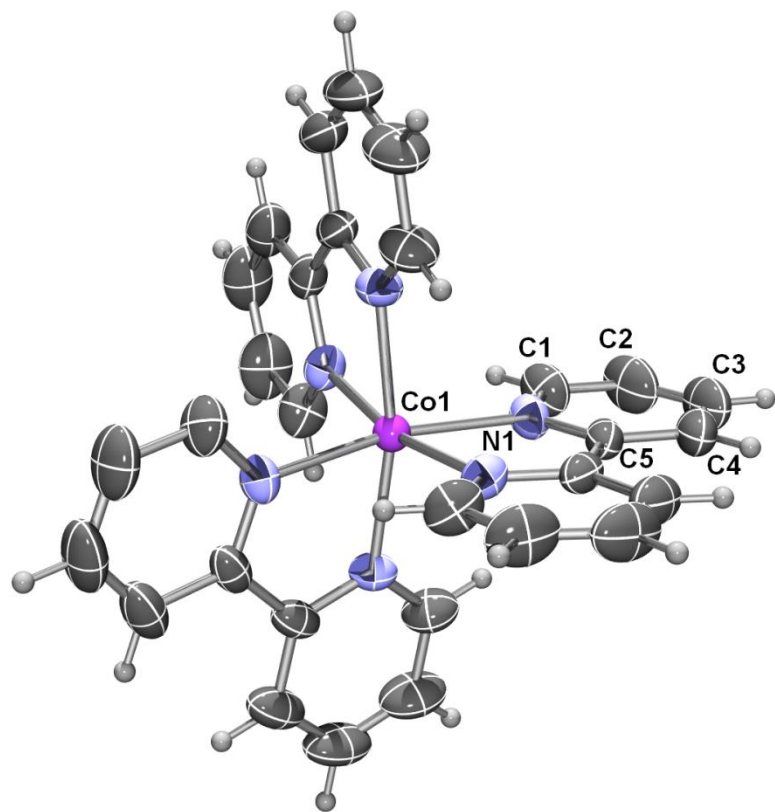
P



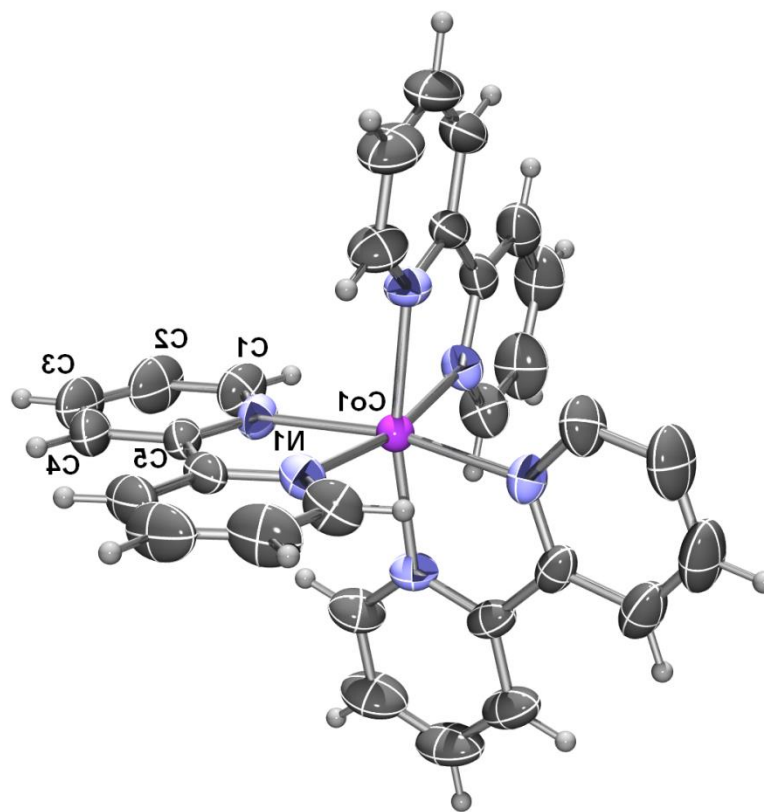
Konformacijska kiralnost



Kiralnost na oktaedarski koordiniranom atomu



Λ



Δ